



## Resolution of *trans*-( $\pm$ )-2-(pyrrolidinyl)cyclohexanol and its methyl ether using boric acid and chiral 1,1'-bi-2-naphthol

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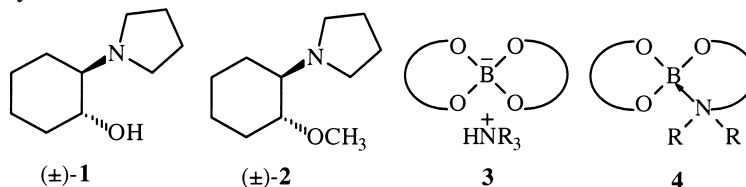
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### Abstract

The *trans*-( $\pm$ )-2-(pyrrolidinyl)cyclohexanol **1**, and its methyl ether **2** were resolved using chiral 1,1'-bi-2-naphthol and boric acid in THF or CH<sub>3</sub>CN. X-Ray structural analysis was carried out for the diastereomeric complex **3** obtained using **2**. © 1999 Elsevier Science Ltd. All rights reserved.

Chiral amino alcohols are important compounds used widely in asymmetric synthesis as chiral ligands or chiral auxiliaries<sup>1</sup> and in medicinal chemistry as therapeutic agents.<sup>2</sup> These compounds are generally prepared from the naturally occurring  $\alpha$ -amino acids. However, they can also be readily prepared through opening of epoxides followed by resolution using tartaric acid<sup>3</sup> or binaphthyl phosphoric acid.<sup>4</sup>

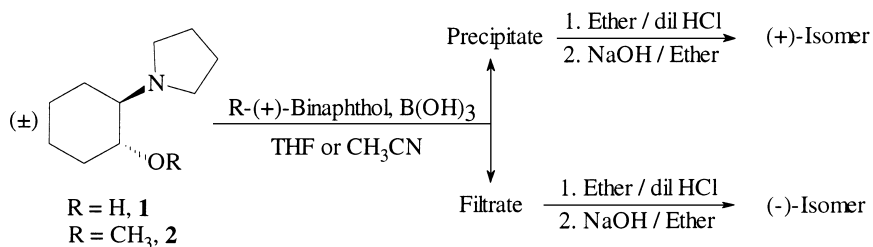
During the course of our research efforts on the synthesis and application of chiral amino alcohols,<sup>5</sup> we attempted the resolution of the amino alcohol **1**, prepared through opening of cyclohexene oxide with pyrrolidine<sup>6</sup> and its methyl ether **2** using tartaric acid and binaphthyl phosphoric acid, but these experiments did not yield fruitful results.



In recent years, we have developed resolution procedures for 1,1'-bi-2-naphthol, 1,2-diphenyl-1,2-ethanediol and 1,1,2-triphenyl-1,2-ethanediol using (*S*)-proline via preparation of diastereomeric inclusion complexes or borate complexes.<sup>7</sup> It has also been observed that racemic 1,1'-bi-2-naphthol can be readily resolved using chiral  $\alpha$ -methylbenzylamine in the presence of B(OH)<sub>3</sub> in solvents such as CH<sub>3</sub>CN and THF through formation of the corresponding Brønsted acid–ammonium salt **3**.<sup>7</sup> Accordingly, in principle, it should be possible to resolve the racemic amine derivatives through preparation of such diastereomeric salts using chiral 1,1'-bi-2-naphthol; indeed, this was observed. It was found that the

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resolution of **1** and **2** can be readily achieved using chiral 1,1'-bi-2-naphthol and B(OH)<sub>3</sub> in various solvents (Scheme 1). The results are presented in Table 1.



Scheme 1.

When the amino ether **2** and B(OH)<sub>3</sub> were used in a 3:1 ratio (Table 1, entries 7 and 8), (1*S*,2*S*)-**2** was obtained in a higher ee with lower chemical yield from the precipitate fraction. Also the (1*R*,2*R*)-**2** was obtained in a lower ee from the filtrate fraction. Enrichment of the partially resolved amino ether **2** was also carried out following this procedure (Scheme 1). The maximum  $[\alpha]_{\text{D}}^{25}$  value obtained was +41.4. No further enhancement of the  $[\alpha]_{\text{D}}^{25}$  value was obtained when samples with  $[\alpha]_{\text{D}}^{25} = +41.4$  were used.

Table 1  
Resolution of **1** or **2** using (*R*)- or (*S*)-1,1'-bi-2-naphthol and boric acid

S No	Substrate	1,1'-Bi-2-Naphthol R or S	Solvent	Chiral <b>1</b> or <b>2</b> obtained from			
				Precipitate		Filtrate	
	% ee			% ee <sup>a</sup> /Conf.	Yield (%) <sup>b</sup>	% ee <sup>a</sup> /Conf.	Yield (%) <sup>b</sup>
1 <sup>c</sup>	<b>1</b> , 00	<i>S</i>	MeOH	67 (1 <i>R</i> ,2 <i>R</i> )	20	22 (1 <i>S</i> ,2 <i>S</i> )	60
2 <sup>d</sup>	<b>1</b> , 00	<i>S</i>	THF	38 (1 <i>R</i> ,2 <i>R</i> )	40	36 (1 <i>S</i> ,2 <i>S</i> )	45
3 <sup>e</sup>	(1 <i>R</i> ,2 <i>R</i> )- <b>1</b> , 38	<i>S</i>	THF	79 (1 <i>R</i> ,2 <i>R</i> )	40	11 (1 <i>R</i> ,2 <i>R</i> )	55
4 <sup>f</sup>	(1 <i>R</i> ,2 <i>R</i> )- <b>1</b> , 79	<i>S</i>	THF	>99 (1 <i>R</i> ,2 <i>R</i> )	35	08 (1 <i>R</i> ,2 <i>R</i> )	50
5 <sup>g</sup>	<b>1</b> , 00	<i>R</i>	THF	49 (1 <i>S</i> ,2 <i>S</i> )	40	14 (1 <i>R</i> ,2 <i>R</i> )	50
6 <sup>h</sup>	<b>2</b> , 00	<i>R</i>	MeOH	74 (1 <i>S</i> ,2 <i>S</i> )	20	18 (1 <i>R</i> ,2 <i>R</i> )	65
7 <sup>i</sup>	<b>2</b> , 00	<i>R</i>	CH <sub>3</sub> CN	77 (1 <i>S</i> ,2 <i>S</i> )	41	65 (1 <i>R</i> ,2 <i>R</i> )	48
8 <sup>j</sup>	<b>2</b> , 00	<i>R</i>	CH <sub>3</sub> CN	83 (1 <i>S</i> ,2 <i>S</i> )	30	44 (1 <i>R</i> ,2 <i>R</i> )	63
9 <sup>k</sup>	(1 <i>S</i> ,2 <i>S</i> )- <b>2</b> , 83	<i>R</i>	CH <sub>3</sub> CN	>99 (1 <i>S</i> ,2 <i>S</i> )	42	66 (1 <i>S</i> ,2 <i>S</i> )	42
10 <sup>l</sup>	(1 <i>R</i> ,2 <i>R</i> )- <b>2</b> , 67	<i>S</i>	CH <sub>3</sub> CN	97 (1 <i>R</i> ,2 <i>R</i> )	43	37 (1 <i>R</i> ,2 <i>R</i> )	52

a. Based on the maximum  $[\alpha]_{\text{D}}^{25}$  value +41.4 observed for **2** (**1** was converted to **2** and the  $[\alpha]_{\text{D}}^{25}$  was used to determine the ee). b. Yields of isolated products. c. (±)-**1** (10 mmol), B(OH)<sub>3</sub> (5 mmol) and (*S*)-(-)-1,1'-bi-2-naphthol (10 mmol) in MeOH (20 ml) were stirred for 6 hr at rt. d. (±)-**1** (125 mmol), B(OH)<sub>3</sub> (62.5 mmol) and (*S*)-(-)-1,1'-bi-2-naphthol (125 mmol) in THF (125 ml) were stirred for 6 hr at rt. e. (1*R*,2*R*)-**1** (49.4 mmol, 38% ee), B(OH)<sub>3</sub> (24.7 mmol) and (*S*)-(-)-1,1'-bi-2-naphthol (49.4 mmol) in THF (50 ml) were stirred for 6 hr at rt. f. (1*R*,2*R*)-**1** (17 mmol, 79% ee), B(OH)<sub>3</sub> (8.5 mmol) and (*S*)-(-)-1,1'-bi-2-naphthol (17 mmol) in THF (30 ml) were stirred for 6 hr at rt. g. (±)-**1** (10 mmol), B(OH)<sub>3</sub> (5 mmol) and (*R*)-(+)-1,1'-bi-2-naphthol (10 mmol) in THF (20 ml) were stirred for 6 hr at rt. h. (±)-**2** (10 mmol), B(OH)<sub>3</sub> (5 mmol) and (*R*)-(+)-1,1'-bi-2-naphthol (10 mmol) in MeOH (40 ml) were stirred for 12 hr at rt. i. (±)-**2** (19 mmol), B(OH)<sub>3</sub> (9.5 mmol) and (*R*)-(+)-1,1'-bi-2-naphthol (19 mmol) in CH<sub>3</sub>CN (70 ml) were refluxed for 12 hr. j. (±)-**2** (60 mmol), B(OH)<sub>3</sub> (20 mmol) and (*R*)-(+)-1,1'-bi-2-naphthol (40 mmol) in CH<sub>3</sub>CN (150 ml) were refluxed for 12 hr. k. (1*S*,2*S*)-**2** (18.2 mmol, 83% ee), B(OH)<sub>3</sub> (9.1 mmol) and (*R*)-(+)-1,1'-bi-2-naphthol (18.2 mmol) in CH<sub>3</sub>CN (80 ml) were refluxed for 12 hr. l. (1*R*,2*R*)-**2** (10 mmol, 67% ee), B(OH)<sub>3</sub> (5 mmol) and (*S*)-(-)-1,1'-bi-2-naphthol (10 mmol) in CH<sub>3</sub>CN (40 ml) were refluxed for 12 hr.

The enantiomeric excess of the amino ether **2** with  $[\alpha]_D^{25} = +41.4$  was analysed to be >99% by  $^1\text{H}$  NMR (400 MHz) analysis in  $\text{CDCl}_3$  using (*S*)-(+)-binaphthyl-2,2'-diyl phosphoric acid.<sup>8</sup>

X-Ray structural analysis<sup>9</sup> of the crystal of the complex obtained in the amino ether resolution experiment (Table 1, entry 8) was carried out. It revealed that the 1,1'-bi-2-naphthol forms a Brønsted acid with boric acid, which in turn gives diastereomeric salt **3** with the amino ether **2**. The complex **3** crystallised along with three  $\text{CH}_3\text{CN}$  molecules. The ORTEP diagram of the crystal structure is shown in Fig. 1.

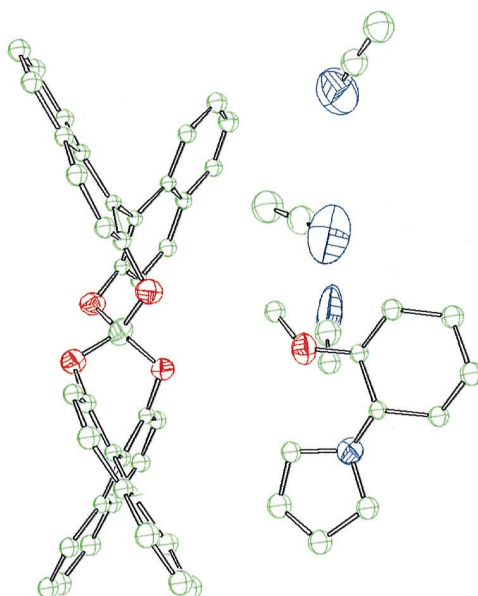


Figure 1. ORTEP diagram for complex **3**

Suitable crystals for X-ray analysis could not be obtained in the case of the diastereomeric complex prepared using amino alcohol **1**. In this case, there is a possibility of the formation of a Lewis acid complex of the type **4**.<sup>10</sup> However, the IR spectrum of the diastereomeric complex obtained from **1** exhibited an OH absorption at  $3410\text{ cm}^{-1}$ , indicating that the complex formed here could also be the Brønsted acid complex **3**.

Experimental procedure for the resolution of **2**: (*R*)-(+)-1,1'-Bi-2-naphthol (40 mmol, 11.45 g),  $\text{B}(\text{OH})_3$  (20 mmol, 1.24 g) and the racemic amino ether **2** (60 mmol, 11 g) were refluxed in  $\text{CH}_3\text{CN}$  (150 ml) for 12 h. The reaction mixture was cooled to room temperature and filtered. The precipitate was suspended in a mixture of ether (80 ml) and dil. HCl (1N, 80 ml) and stirred until complete dissolution occurred. The (*R*)-(+)-1,1'-bi-2-naphthol was recovered from the ether layer. The aqueous layer was treated with NaOH/ether and the free amino ether was extracted with ether (3×50 ml). The combined organic extracts were washed with saturated brine, dried over anhydrous magnesium sulphate and evaporated to dryness to obtain (1*S*,2*S*)-**2**, 3.3 g, 83% ee (30% yield). The filtrate was concentrated and the residue was taken up in a mixture of ether (80 ml) and dil. HCl (1N, 80 ml) followed by workup as outlined above to obtain (1*R*,2*R*)-**2**, 6.9 g, 44% ee (63% yield). After further enrichment of the sample with 83% ee following the same procedure, (1*S*,2*S*)-**2** with >99% ee (42% yield) was obtained from the precipitate fraction (Table 1, entry 8). The filtrate fraction after workup gave the amino ether (1*S*,2*S*)-**2**, with 67% ee (42% yield).

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9. Crystal structure analysis: The X-ray diffraction measurements were carried out at 293 K on an automated Enraf–Nonius MACH 3 diffractometer using graphite monochromated Mo-K $\alpha$  ( $\lambda=0.71073$  Å) radiation. Intensity data were collected by the  $\omega$ -scan mode. The data were reduced using the XTAL programme. No absorption correction was applied. The  $\theta$  range for data collection is 1.54 to 24.97°. Crystal structure data: empirical formula C<sub>57</sub>H<sub>54</sub>BN<sub>4</sub>O<sub>5</sub>, colourless rectangular prism (0.3×0.3×0.4 mm), crystal system is orthorhombic, space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>, unit cell dimensions:  $a=11.343(4)$  Å,  $b=16.169(6)$  Å,  $c=26.395(4)$  Å. Volume 4841(2) Å<sup>3</sup>,  $Z=4$ ,  $D_{\text{calc}}=1.216$  Mg/m<sup>3</sup>, absorption coefficient is 0.077 mm<sup>-1</sup>,  $F(000)=1876$ , index ranges  $0\leq h\leq 13$ ,  $0\leq k\leq 19$ ,  $0\leq l\leq 31$ , total reflections collected were 4823 out of which 4736 were independent reflections with  $R(\text{int})=0.0069$  and  $R(\text{sigma})=0.0941$ . The structure was solved by direct methods and refined by a full-matrix least-squares procedure using the SHELX 86 and SHELX 97 programme package, respectively. The refinement was carried out using 4736 observed [ $F>4\sigma(F)$ ] reflections and converged to a final  $R1=0.0655$ ,  $wR2=0.1268$  and goodness of fit was 1.013 with the largest difference between peak and hole 0.231 and  $-0.196$  e Å<sup>-3</sup>, respectively. Detailed X-ray crystallographic data are available from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 1EZ, UK.
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